

## CHEMICAL MANUFACTURING ACTIVITY

# **Polymers**

# **Polymer Two**

## **Questions**

- ■How do polymers behave?
- ■Do they have the same properties?

# **Materials**

- ■White glue
- ■Borax
- ■Water
- ■Spoon or popsicle stick to stir
- ■Small paper cups
- ■Food coloring
- Graduated cylinder
- ■Ruler
- ■Sealable plastic sandwich bags

### Treparation

■Your teacher may have pre-prepared a borax solution. If not, prepare a borax solution: about 6 mL of borax to 235 mL of water.

#### ✓ Procedure

- 1. Use the ruler to measure and mark 1 cm from the bottom of the small paper cup.
- 2. Add white glue to the 1 cm mark.
- 3. Add a few drops of food coloring and mix.
- 4. Measure 7 mL of water in the graduated cylinder and add to the glue. Mix well and pour into a plastic bag.
- 5. Measure 8 mL of the borax solution using the graduated cylinder and add it to the glue solution in the plastic bag. Mix well by kneading.
- 6. If it is too sticky, add borax solution one drop at a time. If it is to stringy, add glue one drop at a time.
- 7. Once the polymer is formed, you may remove it from the cup and knead it.
- 8. Pull your polymer apart, string it out, twist it, and roll it into a ball.
- 9. Write your observations about your polymer.

#### \*\* Conclusions

1. What happened when you combined the glue solution and the borax solution?

2. Explain how this is a polymer.